



#### **Sustainability Teaching Unit**

# Worksheets for use in elementary school science classes

These worksheets are based on a one-week research course for elementary school students, which is part of the *Forscherwelt* or *Research World* education initiative.

The teaching concept and program were developed under the guidance of Prof. Dr. Katrin Sommer, Chair of Chemistry Didactics at Ruhr University Bochum, Germany, with the support of Henkel experts.

The experiments are suitable for third or fourth grade students.





# Symbols used



**Problem/question** 



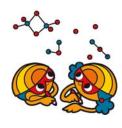
**Assumption** 



**Experiment instructions** 



**Observations** 



**Record your observations** 



**Results** 





#### Saving electricity – saving energy

Worksheet 1

You can save electricity, and therefore energy, at home too.

Pia's father, for example, says that the water you use for cooking pasta boils more quickly if you cover the pot with a lid. He says you use less electricity before the water starts to boil if the pot is covered with a lid than if it is uncovered. Is that right?





# What experiment can you do to find out if what Pia's father says is

Each research group has the following materials available for the

- 1 liter (34 fl. oz.) water in a measuring cup
- 1 large beaker
- 1 hotplate magnetic stirrer
- Aluminum foil
- 1 stopwatch
- 1 thermometer
- 8 boiling stones (which are put into the water to be boiled)

Assignment: Discuss with your partners how you can work with the other groups to answer the question.

#### TIP:

When researchers want to compare something, they often carry out two experiments. The two experiments must be carried out in the same way, the only difference being the specific aspect (in this case with or without lid) you want to investigate.

Write down the <u>method</u> that you and the others plan to use to test the claim made by Pia's father:





#### Saving electricity – saving energy

Worksheet 2

To answer the question being investigated, we will split into two groups.

The first group will heat the water in a container with a lid. This group will be called the **experimental group**.

The other group will heat the water in a container without a lid. This group will be called the **control group**.



Put 1 liter (34 fl. oz.) of water into a large (tall) beaker.

Measure the temperature of the water.

Temperature:

The water should be about 20°C/68°F (± 3°C/5°F). Put 8 boiling stones into the water and place it on the hotplate.

# Tick the group to which you belong. Experimental group: Fold a piece of aluminum foil in half so that the foil is doubled. Use the foil as a lid to cover the beaker. Carefully pierce a hole in the center through which you can insert the thermometer. Plug in the hotplate and turn it to the highest setting. Use the stopwatch to time how long it takes for the water to start boiling. Write down how many minutes it takes for the water to reach 40°C (104°F), 60°C (140°F), 80°C (176°F) and finally 99°C (210°F). Enter the figures in the table below.

Control group: Leave the beaker uncovered and carefully put the thermometer into it. Plug in the hotplate and turn it to the highest setting. Use the stopwatch to time how long it takes for the water to start boiling. Write down how many minutes it takes for the water to reach 40°C (104°F), 60°C (140°F), 80°C (176°F) and finally 99°C (210°F). Enter the figures in the table in worksheet 3.





# Saving electricity – saving energy

Worksheet 3

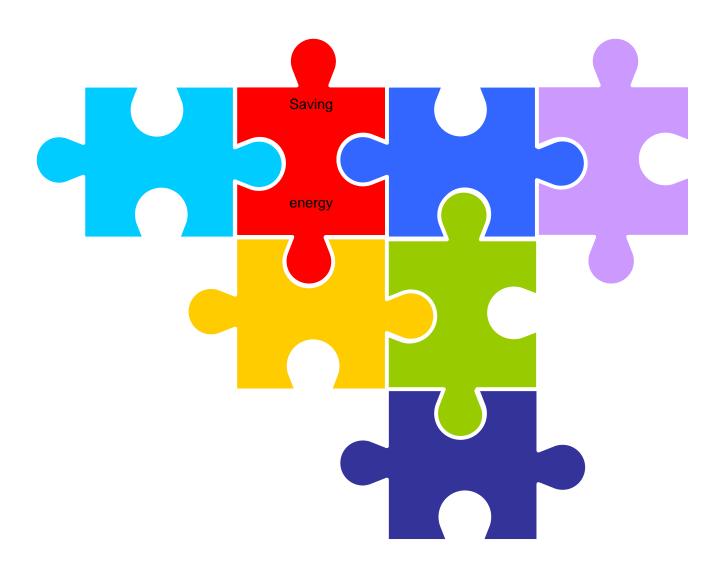
Speak to your neighboring control or experimental group and enter their figures in the table, using a different color.

	Experimental group Time with lid (mm:ss)	Control group Time without lid (mm:ss)
40°C/104°F_ 60°C/140°F		
80°C/176°F_ 99°C/210°F_		



Was Pia's father right? Record the results of your investigations:





# .....using energy efficiently





#### Avoiding and recycling waste

Worksheet 4



How can recycled paper be produced from cardboard packaging?

Waste paper and cardboard are collected in a waste paper bank. The waste paper can be used to produce recycled paper.

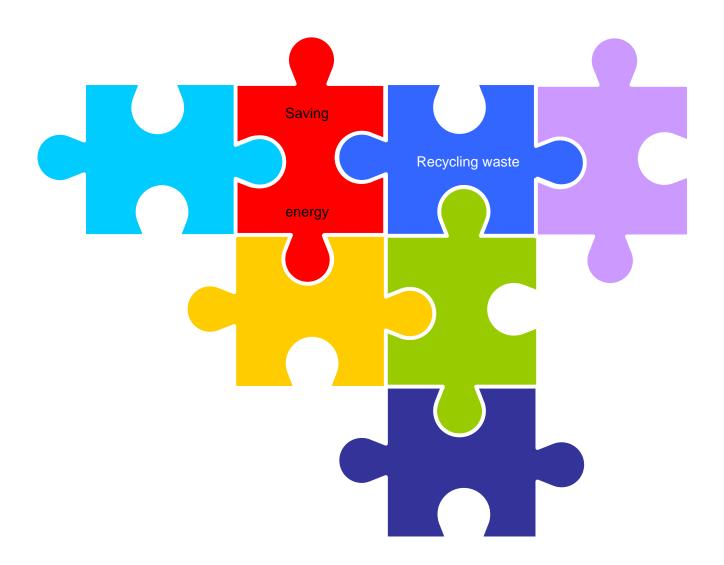


Work with your partner.

- 1. Weigh out 15 g (0.5 oz.) of Persil cardboard. Tear the card into pieces the size of your thumbnail and put the pieces of card into a tall container.
- 2. Use the measuring cylinder to measure 200 ml (7 fl. oz.) of water and add the water to the pieces of card in the tall container.
- 3. Blend the card and water mixture until you have a gray pulp. You should not be able to see any large pieces of card in the pulp.
- 4. Put the pulp into a shallow dish and add 1 liter (34 fl. oz.) of water. Stir the card and water mixture vigorously by hand.
- 5. Use the fly screen as a sieve: One of you stretches the fly screen across the sink and holds it tightly. The other one then pours the pulpy mixture onto the fly screen so that excess water can drip through. When the mixture stops dripping, put the fly screen containing the pulp onto the table, place a piece of fabric on top and turn both over.
- 6. Roll the rolling pin over the screen several times.
- 7. You can then put the piece of fabric with the "paper" onto the drying rack to dry.







.....avoiding, separating and recycling waste







#### What is a raw material?

Pens, paper, T-shirts, toys – virtually everything we come across in our daily lives is produced from a variety of raw materials. Wood is the raw material from which furniture or paper is made, for example, while cotton is the raw material generally used to make T-shirts. Plastic toys are produced from crude oil, another raw material. Raw materials are therefore the basic materials from which products are made for us in one or more steps.

#### What does renewable mean?

Plants grow relatively quickly and are then harvested. At this point, new plants or seeds are planted. They are called "renewable raw materials."

Crude oil, which can be used to produce a whole range of things from gasoline to plastic, does not grow back. Once all the crude oil on Earth has been used, it cannot simply be "planted" again. Crude oil is **not** a renewable raw material. This means we should use it as efficiently and as sparingly as possible.





Worksheet 5

#### Starch

You may have heard the word "starch" before. You sometimes need to use cornstarch for baking, for example. This type of starch looks similar to flour and has similar properties. Flour also contains starch.

It is not just for baking that starch can be used, however. Starch is also used as a basic material (or raw material) for adhesives – for a glue stick, for example.

Where does starch actually come from? From plants? Do you have any idea which plants contain starch?

Today you will learn how to detect the presence of starch.

You can use a substance called Lugol's solution to detect the presence of starch. Lugol's solution is a pink to purple colored liquid that contains iodine. You may be familiar with iodine from its use in medicine. Iodine-containing medicine is used to disinfect a wound, for example. Iodine also has another property, however: It turns dark blue or black when starch is present.



Now conduct an experiment.

- First use a small spatula to put a spatula tip of cornstarch into a test tube.
- Add 2 ml (½ teaspoon) of water and carefully shake the test tube.
- Then add 4 drops of Lugol's solution to the test tube.

Record your observations:		





Prof. Dr. Katrin Sommer, Chair of Chemistry Didactics

#### Renewable raw materials /1

Worksheet 6



#### Which foods contain starch?

Starch is present in a number of foods. You will be shown a selection of different types of food.

First think about which of the foods you are shown could contain starch. Record your assumptions:





Use Lugol's solution to find out whether starch is present in various foods. Enter your results in the table.

Food	Instructions	Starch	No starch
Potato	Drizzle 3 drops of Lugol's solution over a slice of potato.		
Cucumber	Cut a slice of cucumber and drizzle 3 drops of Lugol's solution over it.		
Rice	Crush a few grains of rice in a mortar and put them on a watch-glass. Add 3 drops of Lugol's solution.		
Corn	Crush a few corn kernels in a mortar and put them on a watch-glass. Add 3 drops of Lugol's solution.		





Worksheet 7

#### Obtaining starch from food



You have now learned that starch is present in potatoes, wheat, rice and corn. To use this starch to make an adhesive, you must first find a way of getting the starch out of the food.



Here are the instructions you need to conduct the experiment – but somehow they have gotten mixed up. First put the sentences below in the correct order. Then cut out the individual boxes and glue them in the right order into your notebook or onto a new sheet of paper in your science folder.

Put the remaining mixture back into the first bowl and repeat steps two and three, but using only 200 ml (7 fl. oz.) of water. Wait five minutes and then carefully

strain off the liquid. Leave the white residue at the bottom in the bowl.

Put the residue into a dish and place the dish in the oven at 180°C (350°F) for 20

Add 300 ml (10 fl. oz.) of water to the grated food in the plastic bowl and stir with

Change and of the foods (2.6 notations or 150 g./F. o., ) of commonly and grate if

Choose one of the foods (3-6 potatoes or 150 g (5 oz.) of cornmeal) and grate if necessary (into a plastic bowl).

Put a dish towel above a second plastic bowl, pour in the mixture and squeeze out the water (liquid). Collect this liquid in a bowl and wait until some sediment settles at the bottom.





Worksheet 8

#### Starch paste containing soap

Soap is said to make a starch paste more solid. Try it out!



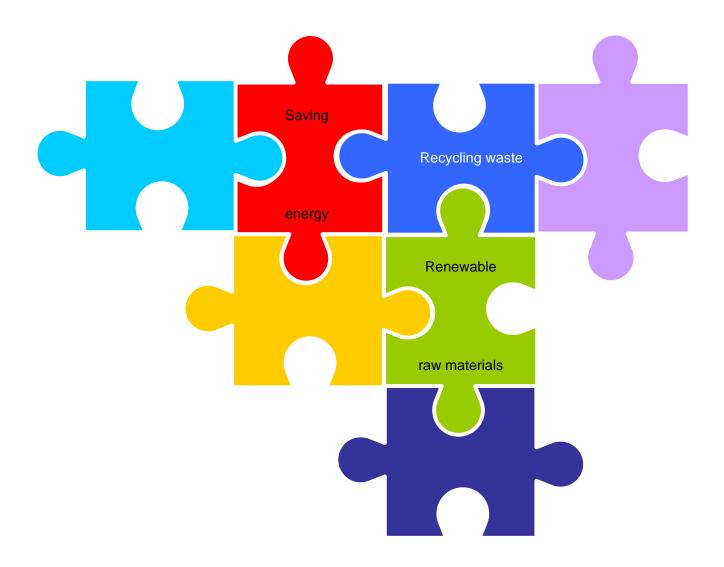
- 1. Use a potato grater to grate about a quarter of a bar of soap.
- 2. In a 150 ml (5 fl. oz.) beaker, dissolve 1 g (¼ teaspoon) of the grated soap in 14 ml (1 tablespoon) of water as thoroughly as possible; this will produce a lather.
- 3. Add 4 g (1 teaspoon) of starch to the lather mixture produced and mix well with the glass rod.
- 4. Heat the mixture on a hotplate to a temperature of about 75°C (167°F), stirring occasionally with the glass rod.
- 5. Repeat steps 2) to 4) using 2 g (½ teaspoon), 3 g (¾ teaspoon) and 4 g (1 teaspoon) of soap.

Which of the 4 glue stick substances is most similar to the original glue stick? How much soap, water and starch did you use for this sample?

Write down the recipe (also known as the formulation):				







.....using renewable raw materials





#### Protecting our health /1

Worksheet 9

Healthy teeth are attractive ©

Bad teeth don't just look bad; they can also cause other diseases.

Taking good care of your teeth protects your health – and gives you a dazzling smile!



#### What causes bad teeth? Sugar?

No – not the sugar itself. Small organisms in our mouth, bacteria, turn sugar into acid. The acid attacks our teeth and leads to tooth decay, also called caries, if we don't clean our teeth well with toothpaste.

#### **Detecting the presence of acids**

**Assignment:** Using a test strip, test water, vinegar, lemon juice, lemonade and lactic acid to find out whether they are acidic. First, guess which liquid is the most acidic, second most acidic, etc. and write down the order on the left of the table below.

- 1. Write the names of the substances to be tested on the test tubes.
- 2. Put about a three-finger width of each liquid into a test tube.
- 3. Using tweezers, hold an indicator paper in each sample and make a note of the color that becomes visible.
- 4. Also write down the number that has been assigned to this color on the container for the indicator paper.

	Guessed Order	Color	Number	Measured Order	
Tap water: Vinegar: Lemon juice: Lemonade: Lactic acid:					





Prof. Dr. Katrin Sommer, Chair of Chemistry Didactics

# Protecting our health /2

Worksheet 10

Eggshells in acid – eggshells as a model substance for teeth

We will be using eggshells as a model substance for teeth. Just like eggshells, teeth contain calcium. Experiments show that acid attacks eggshells, or rather the calcium contained in the eggshells. Toothpaste protects our teeth from acid.

Caries bacteria in the mouth turn the sugar in our mouths into acid. Cleaning our teeth regularly with toothpaste protects our teeth from acid and thus from caries.



#### What effect does acid have on the eggshell?



Work with your partner.

4	Maigh your aggreball and make a note of its weight (A)
	Weigh your eggshell and make a note of its weight. (A)g
2.	Put the piece of eggshell into a beaker and add enough vinegar to cover the
	eggshell completely. Wait for 15 minutes.
3	Carefully dry the eggshell.
_	, , , , ,
4.	Weigh the dry eggshell again. (B)g
5.	Calculate the difference between the first and second times that you weighed the
	eggshell. TIP: To find the difference, you need to subtract one number from
	another. (A) $g - (B)$ $g = g$
6	
0.	Write down what happened:
	90,00





#### Protecting our health /3

Worksheet 11



Cleaning your teeth regularly with toothpaste protects them from caries.

How can you show that toothpaste protects your teeth from acid?

Work with your partner to devise an experiment with the model substance that you have learned about. You can use toothpaste, vinegar and beakers for your experiment.



#### Sample solution

Work with your partner.

- 1. Divide your egg into two halves by drawing a line in the center.
- 2. Rub toothpaste into one side and wait for three minutes.
- 3. Carefully remove the toothpaste with a piece of paper towel.
- 4. Put the egg carefully into the beaker and add enough vinegar to completely cover the egg.

What do you observe? Record your observations:	•





#### Protecting our health /3

Worksheet 12

Toothpastes contain a substance known as fluoride. Fluoride protects the surface of the teeth from acid. Toothpaste also contains other substances that help to clean your teeth when you brush them.



# Which substances in toothpaste help to remove food and plaque from your teeth?



Investigate whether you can use toothpaste, table salt, lime (calcium carbonate) or detergent to clean a coin.

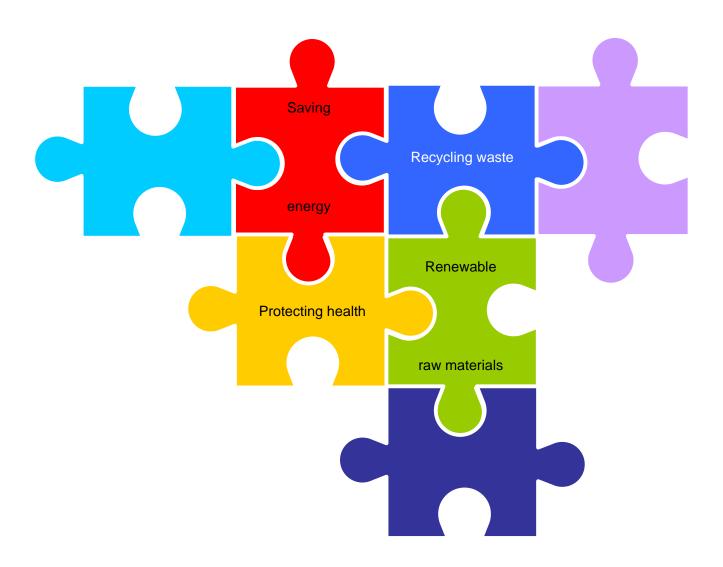
Work in a group of four. Each student receives a different "cleaning agent."

Put a small amount of your cleaning agent onto a cloth and rub the coin for 10 minutes. Compare your coin with the coins cleaned by your neighbors.

What cleaned the coins best and what wasn't as good? Make a note of the order.







.....looking after our health